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## References

- Enraf–Nonius (1988). *CAD-4 VAX/PC Operator's Manual*. Enraf–Nonius, Delft, The Netherlands.
- Gabe, E. J., Le Page, Y., Charland, J.-P., Lee, F. L. & White, P. S. (1989). *J. Appl. Cryst.* **22**, 384–387.
- Hartley, F. R. (1981). *Coord. Chem. Rev.* **35**, 143–209.
- Hori, F., Matsumoto, K., Ooi, S. & Kuroya, H. (1977). *Bull. Chem. Soc. Jpn.* **50**, 138–141.
- Maeda, S., Nishida, Y., Okawa, H. & Kida, S. (1986). *Bull. Chem. Soc. Jpn.* **59**, 2013–2014.
- Newkome, G. R., Frere, Y. A., Fronczek, F. R. & Gupta, V. K. (1985). *Inorg. Chem.* **24**, 1001–1006.
- Newkome, G. R., Puckett, W. E., Kiefer, G. E., Gupta, V. K., Fronczek, F. R., Pantaleo, D. C., McClure, G. L., Simpson, J. B. & Deutsch, W. A. (1985). *Inorg. Chem.* **24**, 811–826.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Sheldrick, G. M. (1990a). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1990b). *SHELXTL/PC. Structure Determination Software Programs*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siedle, A. R. (1981a). *J. Organomet. Chem.* **208**, 115–123.
- Siedle, A. R. (1981b). *Inorg. Chem.* **20**, 1318–1320.
- Siedle, A. R. & Newmark, R. A. (1981). *J. Am. Chem. Soc.* **103**, 1240–1241.
- Siedle, A. R. & Pignolet, L. H. (1982). *Inorg. Chem.* **21**, 135–141.
- Siedle, A. R., Sperl, P. M. & Rusch, T. W. (1980). *Appl. Surf. Sci.* **6**, 149–160.

*Acta Cryst.* (1997). **C53**, 1220–1221

## A Linear Cluster with Mixed Ligands, [WCu<sub>2</sub>S<sub>4</sub>(tpt)<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>]

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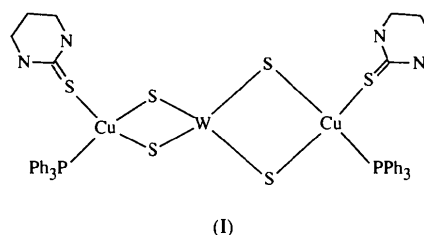
## Abstract

The structure determination of tetra- $\mu$ -sulfido-1:2 $\kappa^4$ S;-1:3 $\kappa^4$ S-bis(tetrahydropyrimidine-2-thione)-2 $\kappa$ S,3 $\kappa$ S-bis-(triphenylphosphine)-2 $\kappa$ P,3 $\kappa$ P-dicoppertungsten, [Cu<sub>2</sub>W-

S<sub>4</sub>(C<sub>4</sub>H<sub>8</sub>N<sub>2</sub>S)<sub>2</sub>(C<sub>18</sub>H<sub>15</sub>P)<sub>2</sub>], is reported. The compound contains a linear cluster core [CuS<sub>2</sub>WS<sub>2</sub>Cu]. Each Cu atom has a distorted tetrahedral coordination, from two S atoms of a tetradentate WS<sub>4</sub><sup>2-</sup> moiety, one S atom of tetrahydropyrimidine-2-thione (tpt) and one P atom of PPh<sub>3</sub>.

## Comment

Several linear clusters MS<sub>4</sub>M'(PPh<sub>3</sub>)<sub>3</sub>·0.8CH<sub>2</sub>Cl<sub>2</sub> (M = Mo, W; M' = Cu, Ag) have been prepared over the last two decades (Müller, Bögge & Schimanski, 1983). Additionally, the linear heterometallic trinuclear clusters [Et<sub>4</sub>N][(PPh<sub>3</sub>)<sub>2</sub>AgS<sub>2</sub>MS<sub>2</sub>Cu(CN)] (M = Mo, W) have been synthesized in recent years (Du, Zhu, Chen, Wu & Lu, 1992a,b). The title compound, (I), also has a linear core [CuS<sub>2</sub>W<sub>2</sub>Cu], but in which both Cu atoms are tetrahedrally coordinated by mixed ligands.



As shown in Fig. 1, the W atom has tetrahedral coordination, WS<sub>4</sub><sup>2-</sup>. Furthermore, each Cu atom is coordinated by a distorted tetrahedron of two S atoms of the tetradentate WS<sub>4</sub><sup>2-</sup> moiety, one S atom of tpt and one P atom of PPh<sub>3</sub>. The average W—Cu, W— $\mu$ -S and Cu— $\mu$ -S distances of 2.7525 (8), 2.201 (2) and 2.321 (2) Å, respectively, are comparable with the corresponding values of 2.740 (3), 2.214 (8) and 2.284 (8) Å found in (PPh<sub>3</sub>)<sub>3</sub>WS<sub>4</sub>Cu<sub>2</sub>·0.8CH<sub>2</sub>Cl<sub>2</sub>. The Cu—S<sub>tpt</sub> bond length of 2.363 (3) Å is longer than that of 2.206 (2) Å in [Cu(tpt)<sub>2</sub>Cl] (Bret, Castan & Jugie, 1983).

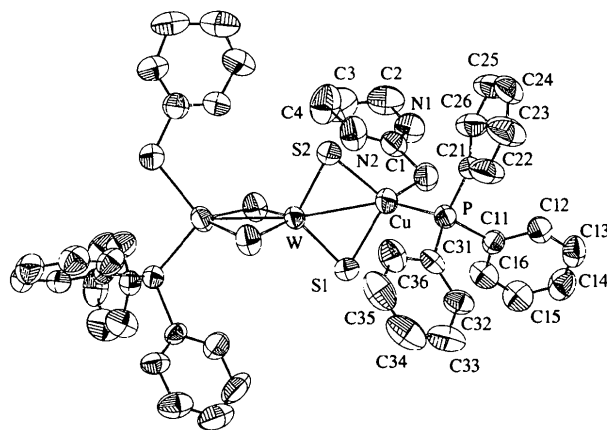


Fig. 1. The molecular structure of (I) showing 50% probability displacement ellipsoids. H atoms have been omitted for clarity.

**Experimental**

The title compound was synthesized by reaction of (PPh<sub>3</sub>)<sub>3</sub>WS<sub>4</sub>Cu<sub>2</sub>·0.8CH<sub>2</sub>Cl<sub>2</sub> with excess tpt in CH<sub>2</sub>Cl<sub>2</sub>. Red-orange air-stable crystals were crystallized by allowing the filtrate to stand for several days after addition of 2-propanol.

*Crystal data*

[Cu<sub>2</sub>WS<sub>4</sub>(C<sub>4</sub>H<sub>8</sub>N<sub>2</sub>S)<sub>2</sub>-(C<sub>18</sub>H<sub>15</sub>P)<sub>2</sub>]  
*M<sub>r</sub>* = 1196.15  
 Monoclinic  
*C2/c*  
*a* = 17.427 (2) Å  
*b* = 10.021 (5) Å  
*c* = 28.639 (3) Å  
 $\beta$  = 106.71 (1)°  
*V* = 4790 (1) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.66 Mg m<sup>-3</sup>  
*D<sub>m</sub>* not measured

Mo *K*α radiation  
 $\lambda$  = 0.71069 Å  
 Cell parameters from 20 reflections  
 $\theta$  = 6.0–27.5°  
 $\mu$  = 3.68 mm<sup>-1</sup>  
*T* = 293 K  
 Block  
 0.40 × 0.40 × 0.20 mm  
 Red–orange

*Data collection*

Rigaku AFC-5R diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction: empirical via  $\psi$  scans (Fair, 1990)  
*T<sub>min</sub>* = 0.196, *T<sub>max</sub>* = 0.479  
 8944 measured reflections  
 4477 independent reflections

2511 reflections with *I* > 3σ(*I*)  
*R<sub>int</sub>* = 0.039  
 $\theta_{\max}$  = 25°  
*h* = 0 → 20  
*k* = 0 → 12  
*l* = -33 → 33  
 3 standard reflections every 300 reflections  
 intensity decay: none

*Refinement*

Refinement on *F*  
*R* = 0.037  
*wR* = 0.037  
*S* = 1.04  
 2511 reflections  
 267 parameters  
 H atoms not refined  
*w* = 1/σ<sup>2</sup>(*F*)

( $\Delta/\sigma$ )<sub>max</sub> = 0.002  
 $\Delta\rho_{\max}$  = 0.35 e Å<sup>-3</sup>  
 $\Delta\rho_{\min}$  = -0.08 e Å<sup>-3</sup>  
 Extinction correction: none  
 Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$B_{\text{eq}} = (8\pi^2/3) \sum_i \sum_j U^{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

|     | <i>x</i>    | <i>y</i>    | <i>z</i>    | <i>B<sub>eq</sub></i> |
|-----|-------------|-------------|-------------|-----------------------|
| W   | 1/2         | 0.05247 (5) | 1/4         | 3.565 (9)             |
| Cu  | 0.37341 (6) | 0.0787 (1)  | 0.16695 (3) | 4.51 (2)              |
| S   | 0.2629 (1)  | 0.2243 (3)  | 0.15577 (8) | 5.50 (6)              |
| S1  | 0.3919 (1)  | -0.0711 (2) | 0.23179 (7) | 5.07 (5)              |
| S2  | 0.4978 (1)  | 0.1784 (2)  | 0.18617 (7) | 4.94 (5)              |
| P   | 0.3360 (1)  | -0.0457 (2) | 0.09829 (7) | 3.74 (4)              |
| N1  | 0.2350 (4)  | 0.4622 (8)  | 0.1859 (3)  | 6.6 (2)               |
| N2  | 0.3615 (5)  | 0.3916 (8)  | 0.2147 (3)  | 7.1 (2)               |
| C1  | 0.2885 (5)  | 0.3682 (9)  | 0.1883 (3)  | 5.0 (2)               |
| C2  | 0.2506 (6)  | 0.5901 (9)  | 0.2123 (3)  | 7.7 (3)               |
| C3  | 0.3240 (6)  | 0.581 (1)   | 0.2536 (4)  | 9.0 (3)               |
| C4  | 0.3879 (6)  | 0.516 (1)   | 0.2422 (4)  | 9.1 (3)               |
| C11 | 0.2290 (4)  | -0.0808 (7) | 0.0806 (3)  | 3.9 (2)               |

|     |            |             |             |         |
|-----|------------|-------------|-------------|---------|
| C12 | 0.1789 (4) | -0.0749 (8) | 0.0324 (3)  | 4.4 (2) |
| C13 | 0.0986 (5) | -0.1065 (9) | 0.0219 (3)  | 5.5 (2) |
| C14 | 0.0678 (5) | -0.145 (1)  | 0.0588 (3)  | 6.3 (3) |
| C15 | 0.1150 (5) | -0.150 (1)  | 0.1059 (3)  | 6.1 (2) |
| C16 | 0.1941 (4) | -0.1163 (9) | 0.1166 (3)  | 5.1 (2) |
| C21 | 0.3551 (4) | 0.0222 (8)  | 0.0442 (2)  | 3.8 (2) |
| C22 | 0.3765 (6) | -0.055 (1)  | 0.0106 (3)  | 6.5 (2) |
| C23 | 0.3900 (6) | 0.004 (1)   | -0.0306 (3) | 8.2 (3) |
| C24 | 0.3813 (5) | 0.137 (1)   | -0.0378 (3) | 6.9 (3) |
| C25 | 0.3610 (5) | 0.2152 (9)  | -0.0048 (3) | 7.0 (3) |
| C26 | 0.3484 (5) | 0.1585 (9)  | 0.0365 (3)  | 5.8 (2) |
| C31 | 0.3783 (4) | -0.2138 (7) | 0.1048 (2)  | 3.7 (2) |
| C32 | 0.3320 (5) | -0.3278 (8) | 0.0946 (3)  | 4.8 (2) |
| C33 | 0.3657 (5) | -0.4534 (9) | 0.1033 (3)  | 6.4 (2) |
| C34 | 0.4479 (5) | -0.464 (1)  | 0.1211 (3)  | 7.5 (3) |
| C35 | 0.4963 (5) | -0.353 (1)  | 0.1300 (3)  | 7.4 (3) |
| C36 | 0.4611 (5) | -0.226 (1)  | 0.1228 (3)  | 6.0 (2) |

Table 2. Selected geometric parameters (Å, °)

|                      |            |          |            |
|----------------------|------------|----------|------------|
| W—Cu                 | 2.7525 (8) | Cu—S1    | 2.337 (2)  |
| W—S1                 | 2.188 (2)  | Cu—S2    | 2.305 (2)  |
| W—S2                 | 2.213 (2)  | Cu—P     | 2.261 (2)  |
| Cu—S                 | 2.363 (3)  | S—C1     | 1.705 (9)  |
| Cu—W—Cu <sup>i</sup> | 169.03 (4) | W—Cu—S2  | 50.95 (5)  |
| Cu—W—S1              | 55.03 (6)  | W—Cu—P   | 131.15 (7) |
| Cu—W—S1 <sup>i</sup> | 132.94 (6) | S—Cu—S1  | 115.2 (1)  |
| Cu—W—S2              | 54.00 (6)  | S—Cu—S2  | 115.99 (9) |
| Cu—W—S2 <sup>i</sup> | 118.61 (6) | S—Cu—P   | 101.70 (8) |
| S1—W—S1 <sup>i</sup> | 111.08 (9) | S1—Cu—S2 | 100.56 (8) |
| S1—W—S2              | 108.44 (7) | S1—Cu—P  | 106.03 (9) |
| S1—W—S2 <sup>i</sup> | 109.22 (9) | S2—Cu—P  | 117.59 (9) |
| S2—W—S2 <sup>i</sup> | 110.44 (9) | Cu—S—C1  | 111.9 (3)  |
| W—Cu—S               | 126.55 (7) | W—S1—Cu  | 74.85 (8)  |
| W—Cu—S1              | 50.12 (5)  | W—S2—Cu  | 75.05 (8)  |

Symmetry code: (i) 1 - *x*, *y*, ½ - *z*.

The title structure was refined by full-matrix least-squares techniques with anisotropic displacement parameters for non-H atoms. H atoms were located in calculated positions and not refined. Structure solution and refinement were carried out on a COMPAQ PROLINEA 4/50 computer using the *MolEN* (Fair, 1990) program package.

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**References**

- Bret, J.-M., Castan, P. & Jugie, G. (1983). *J. Chem. Soc. Dalton Trans.* pp. 301–304.  
 Du, S.-W., Zhu, N.-Y., Chen, P.-C., Wu, X.-T. & Lu, J.-X. (1992a). *Polyhedron*, **11**, 109–113.  
 Du, S.-W., Zhu, N.-Y., Chen, P.-C., Wu, X.-T. & Lu, J.-X. (1992b). *J. Chem. Soc. Dalton Trans.* pp. 339–344.  
 Fair, C. K. (1990). *MolEN. An Interactive Intelligent System for Crystal Structure Analysis*. Enraf-Nonius, Delft, The Netherlands.  
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.  
 Müller, A., Bögge, H. & Schimanski, U. (1983). *Inorg. Chim. Acta*, **69**, 5–16.